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# **Emission Analysis in Multi-Cylinder SI Engine Using Catalytic Coated Piston**

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**ABSTRACT:** Internal combustion engines generate undesirable emissions during the combustion process. The emissions exhausted into the surroundings pollute the atmosphere and harmful to human beings. The emissions are CO, HC,  $NO_X$ ,  $CO_2$  etc... In the present investigation, an attempt has been made to control the emissions by coating metal oxide over the top of piston. The coating is being done by using vacuum coating unit over the piston top. The results obtained from the experiments using the metal oxide coated over the pistons were analyzed. <sup>[4]</sup> The emission control achieved by adopting this technique was found effective.

**KEYWORDS:** Multi-cylinder, SI Engine, Catalytic converter, Piston.

## I. INTRODUCTION

The strong and stringent air pollution regulations on emission standard provide serious challenges for automobile manufacturers. During the past decades, extensive research has been carried out to decrease the emission of toxic components of combustion gases. Considerable development has been made in fulfilling the standards in the recent years.

The new combustion systems and new technologies contribute to these achievements. In-cylinder treatment can be performed by coating metal oxides on the top of the piston using vacuum coating machine.

#### **II. EXPERIMENTAL SETUP**

Five gas analyzers were used for the measurement of HC, CO,  $NO_X$  and  $CO_2$ . Experiments were initially carried out on the engine in order to provide base line data. The engine was stabilized before taking all measurements subsequently. The experiments were repeated by keeping different catalyst coated filter in the exhaust. A multi cylinder Four Stroke Petrol Engine was used. Engine details are given in Table-1.

Parameter	Details
Engine	Four-Stroke Multi cylinder SI
	Engine
Make	Premier Automobile Limited,
	India
Rated power	7.5 kW
Maximum speed	4500rpm
Bore diameter	68mm

#### **Table -1: Engine Specification**



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Stroke	75mm
Displacement volume	1089cc
Compression ratio	7.3:1
Number of cylinders	Four
Cycle	Four
Cooling	Water
Lubrication	Forced Lubrication
Starting system	Battery Ignition System

The experimental layout is shown in Fig 1.

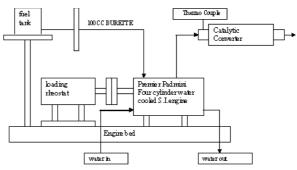


Fig 1: Layout of Test Engine

An electrical Dynamometer is used for loading the engine, the specification which is shown in table-2.

## Table -2: Alternator Specification

Parameter	Details
Loading device	Electrical
Rated power	7kW
Rater speed	1500rpm

Schematic diagram of the Test Engine with Catalytic Coated Pistons are shown in Fig 2 and Fig 3.



Fig 2: Test Engine

Fig3: Pistons Before and After Coating

The present experimental setup for testing the performance of emission of engine shown schematically in figure 2, which consist of multi-cylinder water cooled petrol engine coupled to alternator with water rheostat. The fuel level in



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the fuel tank, flow of cooling water and level of lubricating oil sump is checked before starting the engine. The engine is started by turning the ignition key and the clutch lever is disengaged to couple the engine and adjusted to the speed of 1440 rpm. The engine is allowed to run for 10 minutes to attain steady state condition. At no load condition, CO,  $CO_2$ , HC & NOx are measured by using exhaust gas analyzer. The load on the engine is increased by dipping the electrode in the water of the rheostat after reaching the steady state conditions all the above readings are taken for each increment of the load. The load is completely removed and the speed is reduced before stopping the engine. Then the metal oxide coated piston has been fitted and the emission analysis has been carried out.

## **III. VACUUM COATING MACHINE**

HINDHIVAC Vacuum Coater-12A4D can perform a large number of industrial and laboratory applications like preparation of thui films for Optical and Electronic applications, preparation of specimens or Electron Microscope, etc. The basic unit consists of a cabinet containing a vacuum pumping system together with all the electrical components necessary for the coating process.

#### **3.1 Vacuum Chamber**

The chamber is fabricated from electrochemically polished stainless steel. Three circular glass windows enable visual inspection of the coating process. When the chamber is placed on the base plate it makes a vacuum tight seal with the base plate by means of an 'L' type neoprene gasket.



Fig.4 HINDHIVAC Vacuum Coater

#### **3.2 Measuring Principle**

Measurement was done with the help of Automatic Emission Analyser QRO - 402. (Fig. 5)



Fig. 5 Automatic Exhaust Emission Analyzer

A cooling water pipe line is coiled on the outer wall of the chamber to prevent overheating, and to reduce the out gassing by circulating the water. Alternatively a glass bell jar is supplied along with the unit. The analyzer is



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configured to perform a measurement by applying non-dispersive infrared (NDIR) method for analyzing CO, HC and  $CO_2$  and electro-chemical method for analyzing NO<sub>2</sub> and O<sub>2</sub>. In NDIR analyzing method, an infrared flashing lamp is attached at one end of the sample cell and a detecting sensor at the other end, so that it can detect the component of gas and in turn its density. The electrochemical method measures the gas density by using the quantity of electron which produced in the time of oxidation and reducing reaction of the gas.

#### 3.3 Pumping System

The chamber is evacuated by a HINDHIVAC diff packs pump Model-1 14D and backed by 250 liters per minute, double stage, direct driven, rotary vacuum pump, Model ED-15 with an overload protection.

#### **IV. RESULTS AND DISCUSSION**

#### 4.1 Emission after coating Magnesium Oxide over the Pistons:

The variation in emission of  $NO_X$ ,  $CO_2$ , HC and CO when Ferric Oxide coated catalytic filter used in the exhaust line, is presented in Graphs 1, 2, 3 and 4 respectively. The maximum of the percentage reduction of each pollutant is shown in table 4.

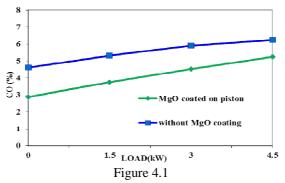
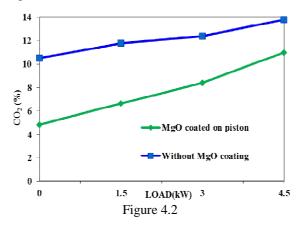


Figure 4.1 shows the variation of CO emission with respect to load behavior base engine i.e without coating on pistons and with MgO catalytic coated pistons. It is observed that the CO emission for the base engine has gradually increased with load and reached 6% CO emission for the maximum load of the investigation. The treatment of magnesium oxide coating on piston has effectively played a vital role in reducing the overall CO emission for the entire load of the investigation. It is observed that the percentage of reduction of emission at minimum load is 37% whereas it is 20% at maximum load. The reason for overall reduction in the emission of CO for all loads is due to the higher combustion temperature in the late combustion phase resulted in the oxidation of CO.

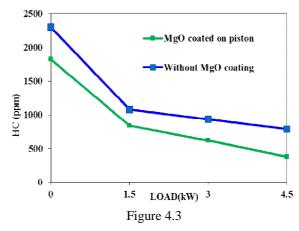




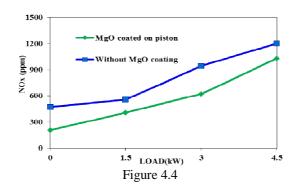
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Figure 4.2 shows the variation of  $CO_2$  emission with respect to load. It is observed that the  $CO_2$  emission has increased with load as observed in the case of CO emission for the both the untreated and treated pistons with MgO. It is noted that the coating of MgO has decreased the overall emission of  $CO_2$  in the exhaust. However the conversion percentage at the minimum load is much higher (54%) than CO emission conversion percentage (37%). But in the higher load the conversion percentage is only 20%, which inferred that at minimum load clean combustion has taken place as the temperature was increased which burn all oxygen content. This increase in temperature is due to the MgO coating on the pistons, whereas at higher load incomplete combustion has taken place and hence the conversion percentage has decreased.



The variation of HC emission with respect to load for the untreated and treated piston head with MgO has been plotted in Figure 4.3. It has been observed that a steep reduction of HC emission in both cases when the load has been increased from minimum to 1.5 kW and there after the emission has remained more or less same for next three loads. Also it is observed that the conversion percentage is 32% in minimum load and it is 51% at maximum load in this study. The tendency of deep decreases till 1.5 kW and the increased conversion efficiency at maximum load is due to the temperature difference occurred for various loads, which leads to change in the combustion efficiency. At higher temperature due to high combustion, the unburnt fuels are less and have the conversion percentage has increased. Referring to the illustration on the emission characteristics with MgO coated piston top, it is seen that both HC and NOx levels are much higher, indicating the effectiveness of the coating.



The variation of  $NO_x$  emission for the both cases is shown in Figure 4.2 d. In this case, it is observed that, the reduction level at lower (minimum) load is 55% and at higher load (4.5 kW) is only 14%. However the overall conversion efficiency has been observed as decreased for all loads. Further, the  $NO_x$  emission level has gradually increased from minimum load to higher load which ensured that the temperature has been increased when the load is increased and hence the increase in temperature during combustion has enhanced the formation of oxides of nitrogen. Hence, from



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this study, it is clear that MgO coating on piston has facilitated this enhancement of emission control of all pollutants of this study to a greater extent.

Table 4		
Pollutants	Maximum Reduction in Percent @ optimum load	
NO <sub>X</sub>	62.59	
СО	26.98	
НС	51.65	
CO <sub>2</sub>	18.77	

Results obtained from the experiments shows that the Magnesium Oxide coated Catalytic filter effectively controls the emission of  $NO_X$  and HC when the engine runs at the optimum load of 4.5 kW also the converter efficiency is more at this optimum load.

#### V. CONCLUSIONS

Following are the conclusions based on experimental results.

- The use of coating Magnesium Oxide over the pistons reduces the emission of
  - NO<sub>x</sub> by 62.59 percent.
  - $CO_2$  by 18.77 percent.
  - CO by 26.98 percent.
  - HC by 51.65 percent.

From the conclusions it is found that the Magnesium Oxide was best catalyst to control the emission from the engines. When this catalyst coating over the piston is used, it may yield better results in reducing the  $NO_x$ , HC and CO.

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